

DECARBOXYLATION OF CYCLIC β -ENAMINOKETOESTERS WITH BORIC ACID

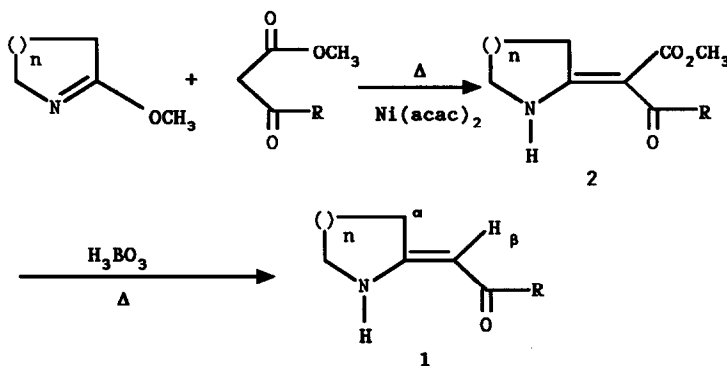
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Summary : *Cyclic β -enaminoketoesters prepared by condensation between lactim ethers and β -ketoesters are decarboxylated without deacylation by thermolysis in presence of boric acid to lead stereospecifically to cyclic β -enaminones.*

Cyclic β -enaminones **1** are important building blocks for synthesis of natural products like ipalbine,¹ myrtine² or methylpelletierine.³ These compounds are unstable and just a few of them have been isolated but often after transformation into β -enaminothioketones⁴. The general method mostly used for the preparation of cyclic β -enaminones involves a sulfur extrusion between a thiolactam and a α -halomethyl ketone as described by Eschenmoser & coll.⁵ but this requires the preparation of such ketones. In connection with our research on the use of boric acid in heterocyclic synthesis, we describe a new stereospecific preparation of cyclic β -enaminones by decarboxylation of cyclic β -enaminoketoesters **2** which are directly prepared⁶ by condensation between lactim ethers and β -ketoesters and can later be decarboxylated to compounds **1**.

Treatment of **2** with sodium alkoxide leads to a β -enaminoester by deacetylation;⁴ drastic acidic media like trifluoroacetic acid⁶ or hydrobromic acid¹ are not regiospecific and methanesulfonic acid⁷ only gives deacylation, which led us to develop less acidic conditions using thermolysis with boric acid.



Heating (1 h, 220°C) **2** with excess solid boric acid leads, after work up (addition of methanol/hydrochloric acid 9/1 then water, neutralization with potassium carbonate, extraction with chloroform and purification), to exocyclic β -enaminones **1** with satisfactory

yields. This decarboxylation is stereospecific and the Z geometry of compounds 1 is unambiguously defined by the chemical shifts of the α -methylene protons, by comparison with the corresponding β -enaminoesters.⁵

Exocyclic β -enaminones 1

n	R	Yield%	Rf (Et ₂ O)	bp °C/mm Hg or mp °C/solvent	H ₃ BO ₃ (eq.)	¹ H-Nmr δ CH ₂ α	δ CH β
1	CH ₃	50	—	122/0.1 ^a	2	2.55	5.10
1	C ₃ H ₇	46	0.32	— ^a	8	2.58	5.07
1	C ₆ H ₅	60	—	109/EtOH ⁹	8	2.73	5.80
2	CH ₃	43	—	98/0.1 ¹⁰	2	2.33	4.93
2	C ₃ H ₇	30	0.44	— ^a	8	2.31	4.82
2	C ₆ H ₅	50	0.51	— ^{a, 10}	10	2.45	5.52
3	CH ₃	39	—	110/0.05	2	2.37	4.95
3	C ₃ H ₇	40	0.64	— ^a	8	2.23	4.88
3	C ₆ H ₅	27	0.70	— ^{a, 10}	10	2.45	5.65

^a Oily products purified by column chromatography (silica gel).

This synthetic pathway provides a useful 3 steps stereospecific method for preparation of β -enaminones from very available starting materials.

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